



**FOUR DECADES IN CHEMISTRY :
THE JOURNEY FROM MCC AND BEYOND**
(A reflection on my personal and professional journey)



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**Frontiers in Chemistry
A Conference to Commemorate the
Birth Centenary of Professor
S.V. Ananthakrishnan
Madras Christian College, Chennai
October 2, 2008**

THE BEGINNING OF A JOURNEY

NEW DELHI TO MADRAS
May 1962



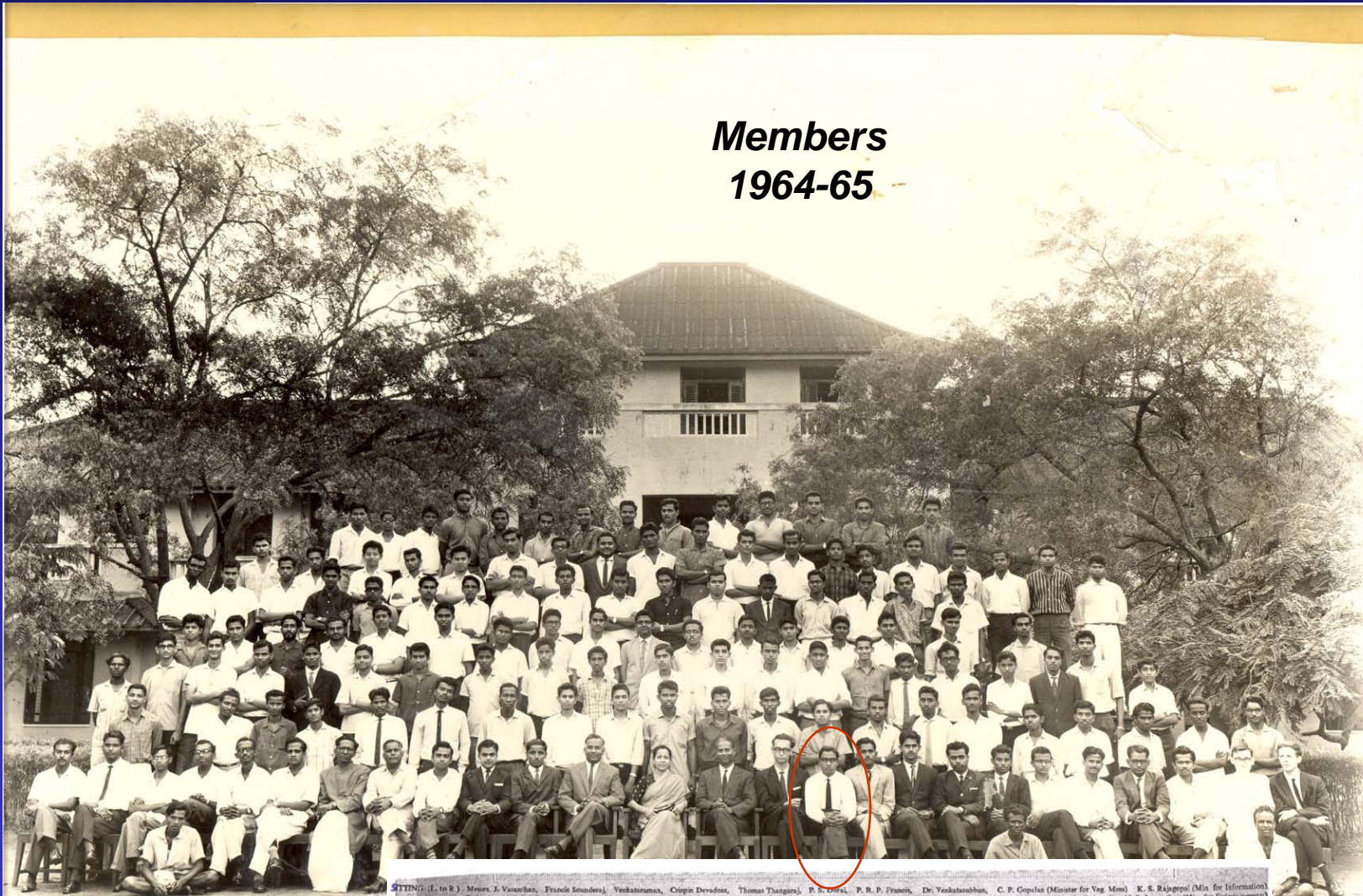
MY JOURNEY TO MADRAS CHRISTIAN COLLEGE

- Finished school at the age of fifteen and a half from the Delhi Higher Secondary Board; Age barred from admission to Delhi University
- Two of my uncles were alumnus of MCC in the late forties
- My father got an introduction to Professor Ananthakrishnan through family friends and sought his advice concerning my education
- I recall my first meeting with Professor Ananthakrishnan at his residence in the Campus along with my father. His advice must have convinced my father that MCC was the right place for me
- My inclination was to study Chemical Engineering ; failed to get admissions in either IIT's or AC College of Technology
- The next best choice was Chemistry !

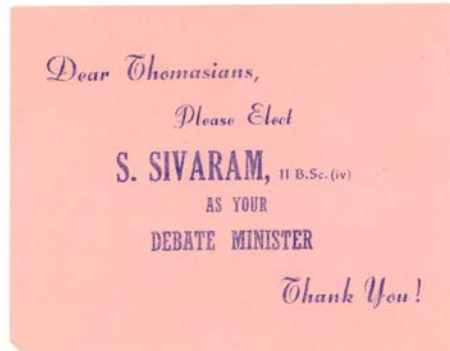


Lesson 1 : We rarely get what we want ; always make the best of second choice

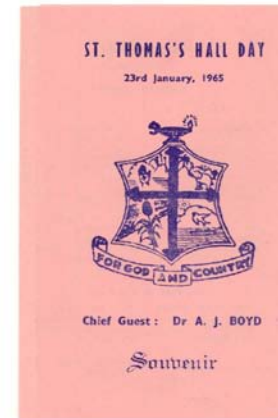
Members 1964-65

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HALL ELECTIONS : FIRST FORAYS IN LEADERSHIP SKILLS



Learning the art of politics and winning popular votes



***Organizing Hall Day
23 January 1965
Chief Guest : Dr A.J.Boyd***

Lesson 2 : Provide early opportunities to learn leadership and organizational skills

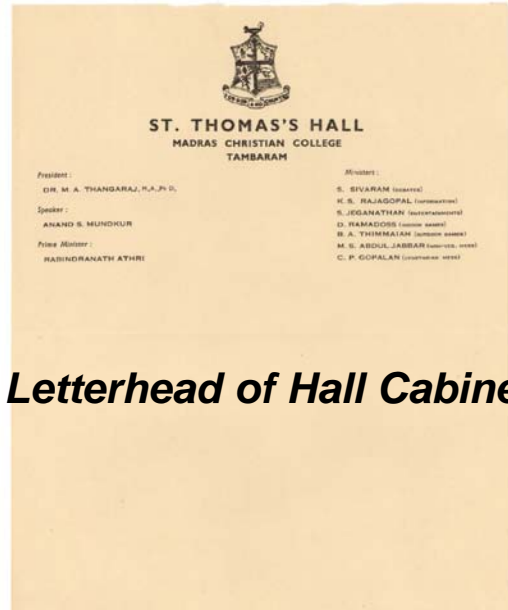
THE PRIDE OF BEING A ST THOMASIAN !

ST. THOMAS'S HALL MADRAS CHRISTIAN COLLEGE, TAMBARAM.

CABINET 1964—65



BENGAL PHOTO STUDIO



Letterhead of Hall Cabinet

**Three years of life
enriching experience !**

Lesson 3 : Learning outside the classroom is more important than inside the classroom

A LETTER FROM DR M.A.THANGARAJ, WARDEN

M. A. THANGARAJ, M.A., Ph.D., (Toronto)
PROFESSOR AND HEAD OF THE DEPT. OF PHYSICS
WARDEN, ST. THOMAS'S HALL

MADRAS CHRISTIAN COLLEGE
TAMBARAM

June 21, 1965

S. Sivaram was a student of this college during the years 1962-65 in the B.Sc. degree class, with Chemistry as his Major Subject and Physics and Mathematics as the Ancillary Subjects. He was a resident member of St. Thomas's Hall. During the three years he was at college.

Sivaram was an active member of the Hall, taking a keen interest in all its activities. He was elected by popular vote to the Hall "Cabinet", as "Minister for Literary Activities and Debates". Himself an able debator, Sivaram represented the Hall in Inter-Hall and Inter-Collegiate Debates and Quiz Competitions. He contributed articles to the Hall Magazine also.

Quiet and unassuming,

SIVARAM SIVARAM SIVARAM
MADRAS

Dr. M. A. Thangaraj, M.A., Ph.D., (Toronto)
PROFESSOR AND HEAD OF THE DEPT. OF PHYSICS
WARDEN, ST. THOMAS'S HALL

Sivaram was a hardworking and sincere student, and was willing to take up responsibilities cheerfully. His character and conduct were very good.

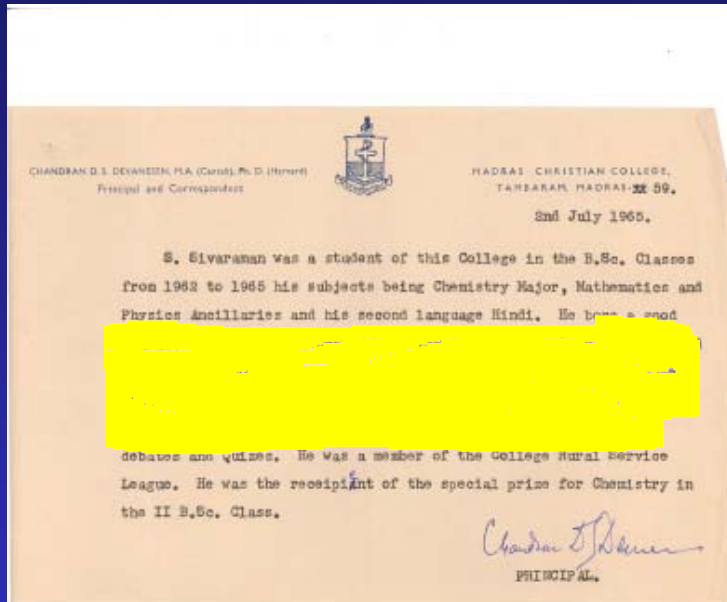
M. A. Thangaraj

"Quiet and unassuming, Sivaram was a hard working and sincere student and was willing to take up responsibilities cheerfully"

Dr M.A.Thangaraj
June 21, 1965

Lesson 4 : Be generous with praise ; It does wonders

A LETTER FROM THE PRINCIPAL

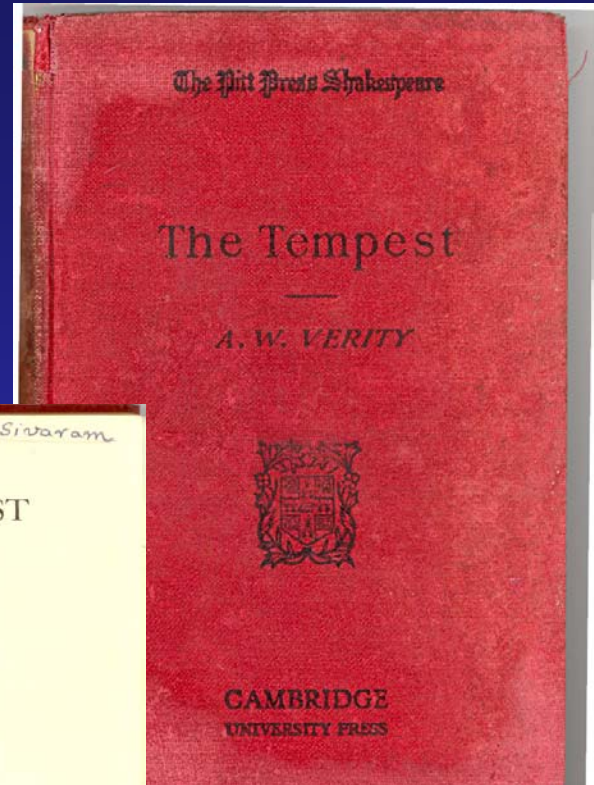
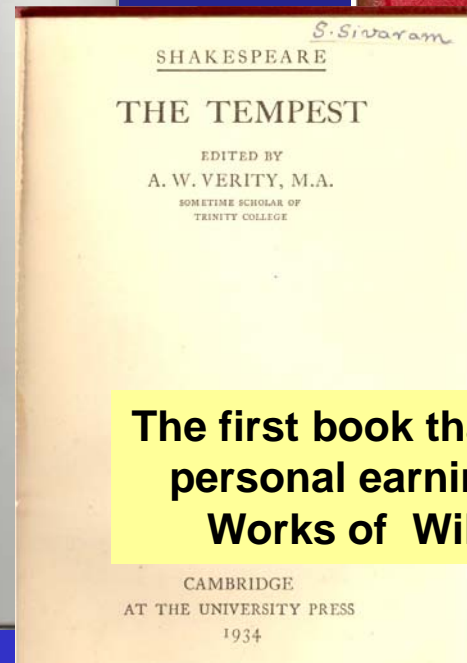
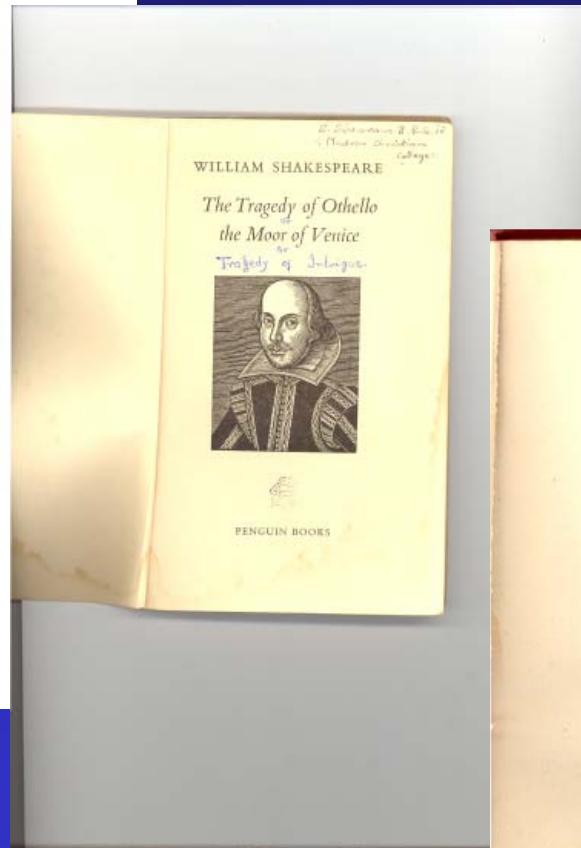
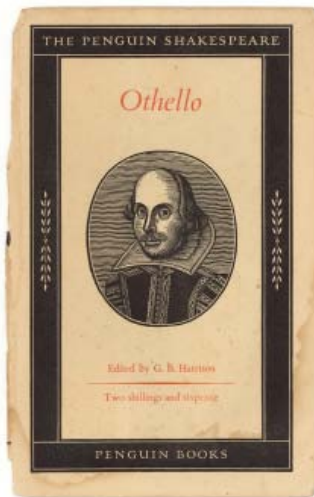


“The record of his class shows that he was a student of more than average ability and a steady worker”

***Dr Chandran Devanesan
July 2.1965***

Lesson 5 : However big you are , show that you care for everyone in your institution

THE BEAUTY OF THE WRITTEN WORD



The first book that I purchased out of my personal earnings was the “Complete Works of William Shakespeare “ !

Lesson 6 : Teach to appreciate the beauty of the written word. Create the love for reading. Language is the window to the soul



THE BEAUTY OF THE WRITTEN WORD

•सिंहासन हिल उठे राजवंशों ने भृकुटी तानी थी,
बूढ़े भारत में आई फिर से नयी जवानी थी,
गुमी हुई आज़ादी की कीमत सबने पहचानी थी,
दूर फिरंगी को करने की सबने मन में ठानी थी।
चमक उठी सन सत्तावन में, वह तलवार पुरानी थी,
बुंदेले हरबोलों के मुँह हमने सुनी कहानी थी,
खूब लड़ी मर्दानी वह तो झाँसी वाली रानी थी॥

**Harivansh Rai
Bacchan**



A life long love affair with books !



**Subadhra Kumari
Chauhan**

मुसलमान औ' हिन्दू है दो, एक, मगर,
उनका प्याला, एक, मगर, उनका
मदिरालय, एक, मगर, उनकी हाला, दोनों
रहते एक न जब तक मस्जिद मन्दिर में
जाते, बैर बढ़ाते मस्जिद मन्दिर मेल
कराती मधुशाला!।५०।

- The Muslim and the Hindu are different, but they drink out of the same cup/ They drink at the same tavern, their wine is also the same/ They remain together so long as they stay away from the temple or mosque/ The temple and the mosque divide but the tavern only unites.



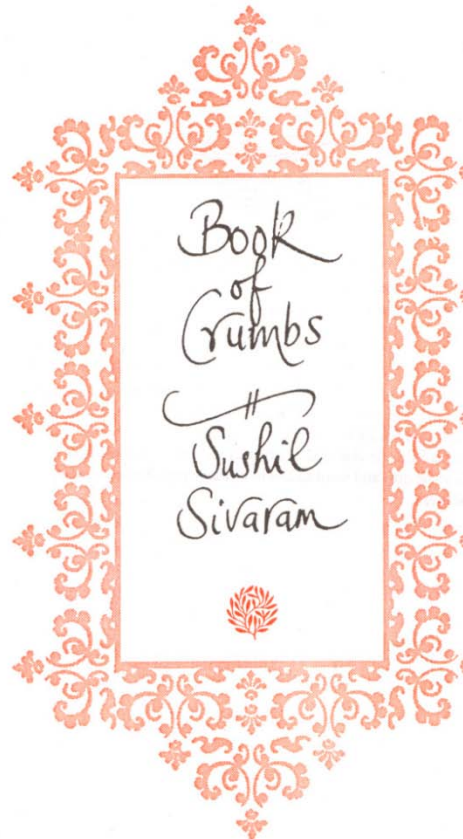
LOVERS OF LITERATURE AND POETRY



**Sushil : A writer
and a poet**



**Rama : Literature, Art
Historian and communicator**



Sivaram ~

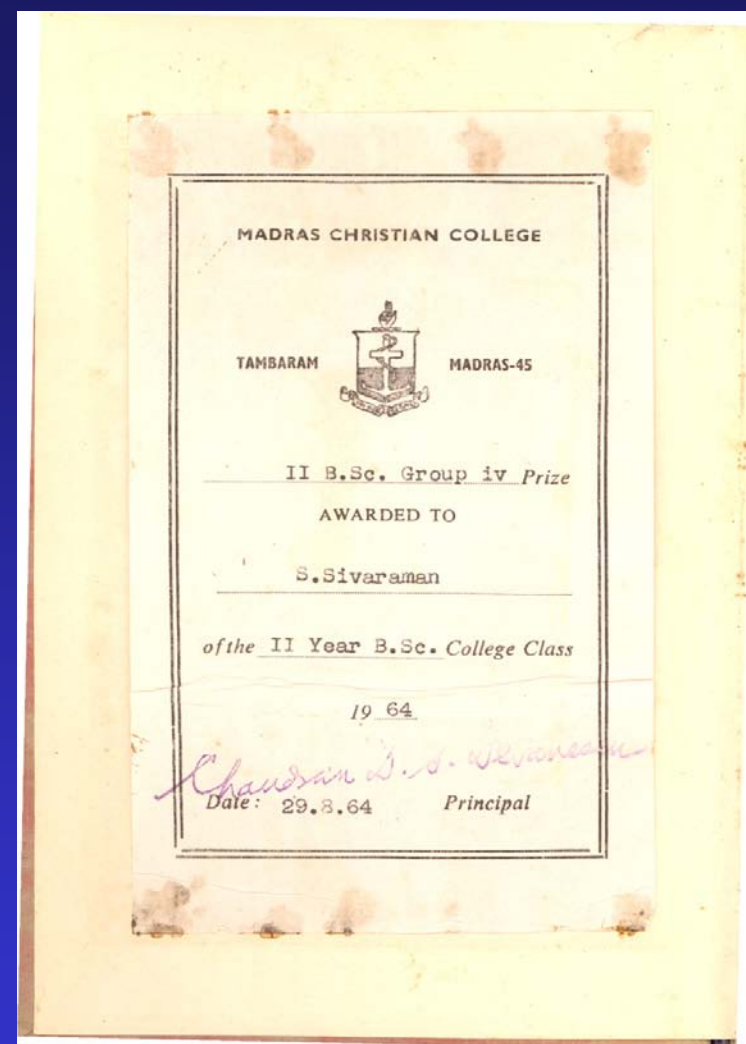
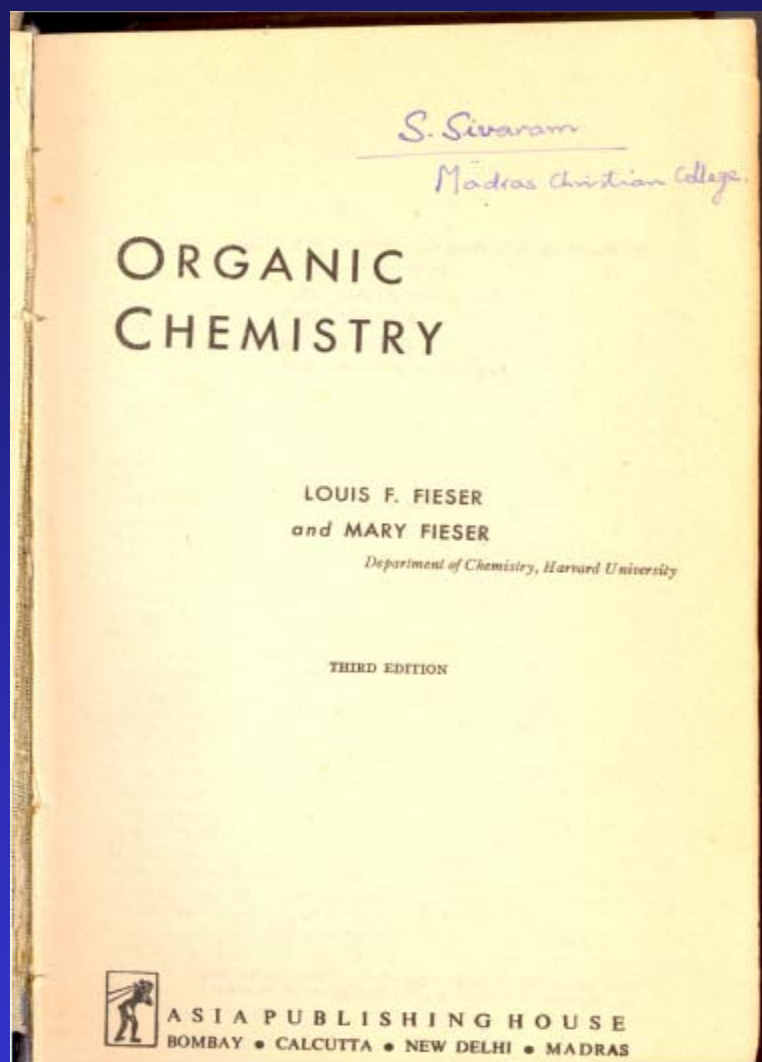
Book of Crumbs ~

Chemistry



You were wedded
To your polycarbonates.
In plastic satin,
Lotus stemmed siphon
Of chemical abstractions.
She was your whore,
A morning song –
Embalmed in turmeric and grey ash scent.
She was a spiritual crank
Of a rusted automobile,
Rotating
Like the little magnetic stirrer
At the bottom of our
Pyrex, conical flask.
Irony sitting pale,
On the Aldrich catalogue,
Schedules painted on the white board,
Teleconferences.
You couldn't hear my tear
Pelting within your outside,
For windows were sealed
And air-conditioners bellowed
Outside your inside.
Aboriginal pamphlets
Of hexagonal carbon double bonds
Like Fractal Geometry,
You left
You're lonely
Patents pending.

MY FIRST POSSESSION IN CHEMISTRY !



Lesson 7 : If you want to find out a person's real aptitude, give him a gift and ask him to select a book from a bookstore

EDUCATION AT MCC

- Committed teachers who were genuinely interested in the student
- An ambience of research and scholarship
- Liberal education; apart from English and Hindi (prose and poetry) I had an opportunity to learn philosophy and economics; a class in moral instruction introduced me to the Old Testament ; led to a life long affair with books
- No early specialization
- A small class of just twenty !
- A compelling urge to excel !

Lesson 8 : Teachers who are committed and show genuine interest in their pupil make good institutions great



THE INEVITABLE EXAMINATIONS !

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THREE-YEAR DEGREE COURSE

B.Sc. DEGREE EXAMINATION, APRIL 1965.

Part III—Branch IV—Chemistry Main—Group A

Paper I

THEORETICAL CHEMISTRY

Time: Three hours.

Maximum: 100 marks.

Answer any FIVE questions.

All questions carry equal marks.

✓1. How was neutron discovered? Write an account of nuclear reactions brought about by neutrons. Explain any one use of a radioactive tracer in physico-chemical investigations.

2. What is meant by (a) molar refraction (b) molar polarization? How are these quantities used in the study of the structure of molecules?

3. How are the following results established:—

- (a) α particles consist of helium nuclei.
- (b) Ordinary water contains a small amount of heavy water.
- (c) Carbon dioxide is a linear molecule?

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THREE-YEAR DEGREE COURSE

B.Sc. DEGREE EXAMINATION, APRIL 1965.

Part III—Branch IV—Chemistry Main—Group A

Paper II

INORGANIC CHEMISTRY

Time: Three hours.

Maximum: 100 marks.

Answer any FIVE questions.

All questions carry equal marks.

1. Describe Morley's experiment for the determination of the composition of water. Under what conditions does water react with the following:—sodium, magnesium, iron, chlorine, sodium peroxide and calcium oxide? Give equations.

✓2. How is pure hydrogen peroxide prepared? Mention its properties and uses. 25 c.c. of acidulated hydrogen peroxide solution required 20 c.c. of decinormal solution of potassium permanganate for complete oxidation. Calculate the volume of oxygen at N.T.P. produced during this reaction.

✓3. How is chlorine manufactured? What are its industrial applications?

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THREE-YEAR DEGREE COURSE

B.Sc. DEGREE EXAMINATION, APRIL 1965.

Part III—Branch IV—Chemistry Main—Group A

Paper III

ORGANIC CHEMISTRY

Time: Three hours.

Maximum: 100 marks.

Answer any FIVE questions.

All questions carry equal marks.

Describe the following syntheses:—

Methyl glyoxal.

Leucine.

Acetonyl acetone.

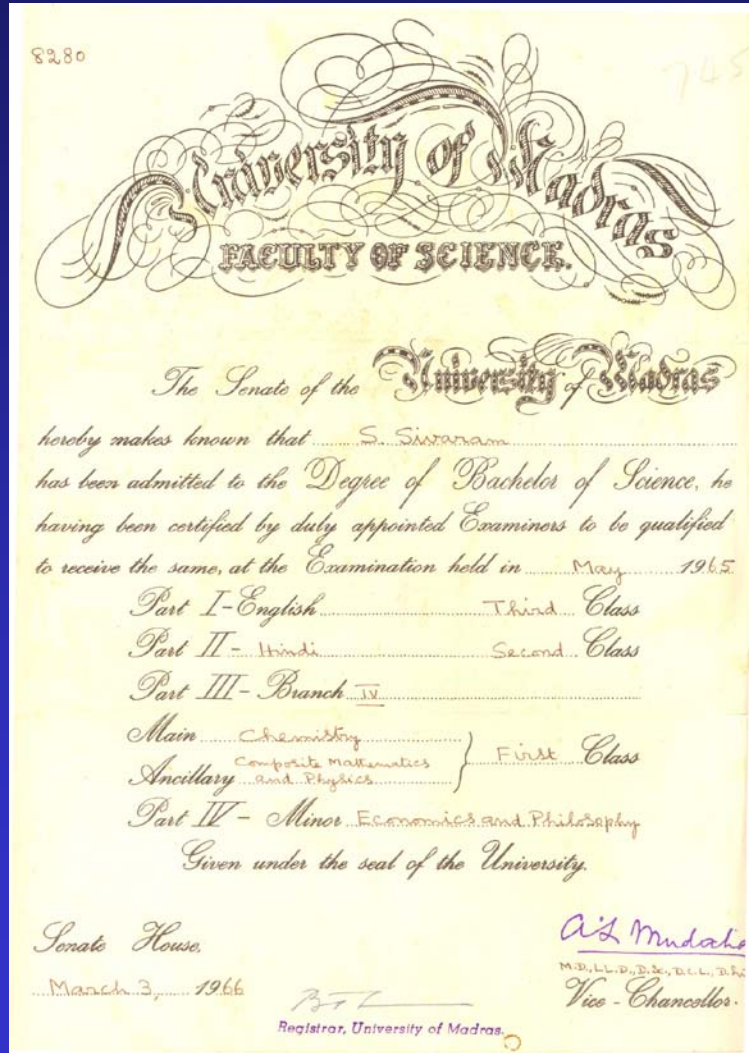
Adipic acid?

Starting from nitrobenzene, how are the following

aniline, azobenzene, benzidine and phenyl-

glycerol synthesised? Describe the action of (a) periodic acid, (b) potassium hydrogen sulphate on glycerol, (c) periodic acid, upon glycerol on heating.

THE THRILL OF A PRIZED AQUISITION



March 3, 1966

- English : Third class
- Hindi : Second class
- Chemistry :First class
- Physics and Mathematics (Ancillary) : First class
- Philosophy and Economics (Minors)

I left MCC, not merely armed with a degree but, with a preparation for life

Lesson 9 : A liberal education is far more important than learning a few subjects; You can rebuild a façade but can lay the foundation only once

IN DISTINGUISHED COMPANY !

***A college which produces such men and women
must have some great genes***

E.C.G. Sudarshan



Prakash Karat



M.V.George



T.N.Seshan



Indra Nooyi



S. Radhakrishnan

TIME TO SAY GOOD BYE



***The class of '65 with
Professor
S.V.Ananthakrishnan***

***A rare photograph of a B. Sc
Class with Professor
Ananthakrishnan !***

***While addressing us at this
valedictory function, SVA in
his characteristic candidness
said that not many of us will
get far ahead in our lives !***



GREATEST REWARDS FOR A STUDENT

K.S. VENKATASUBBAN, M.Sc.,
Lecturer in Chemistry
Madras Christian College
Tambaram East S.O.
MADRAS-79.

June 23, 1967

I have known Shri S. Sivaram for the last three years as a student in the B.Sc. Degree classes. He is one of the brightest students I have come across. He is very intelligent and evinced a keen interest in the subject. As a student he is regular and hardworking.

Pleasant mannered and pleasant, he is courteous and correct in his behaviour towards people. He bears an excellent character and his conduct is very good. I have no doubt that he will give off his best to anything he undertakes. I wish him all success.

K.S. Venkatasubban

(K.S. Venkatasubban)

He is one of the brightest student I have come across. He evinced a keen interest in the subject. I have no doubt he will give his best to anything he undertakes

K.S. Venkatasubban

H. JAYARAMAN, M.A., Ph.D., Dip. German, Dip. French,
Lecturer in Chemistry
Madras Christian College
Tambaram East S.O.
Madras-79.

June 23, 1967.

Mr. S. Sivaram joined this Department for his Bachelor's Degree in Chemistry, in June 1962. He has been a top ranking student in his class and I am confident that he will come out very successful in high First Class in the B.Sc. Degree Examination which he has written just now, and for the which results will be published in July.

I have been teaching Physical Chemistry to him for the past three years and I have been very much impressed by his keen sense of purpose and good understanding of the subject. His laboratory work has been very tidy and his analytical results are regularly accurate.

His conduct and character are excellent and his deportment is exemplary.

I am confident that he will do very well in advanced courses of study in Chemistry or Chemical Technology.

I wish him all success.

H. Jayaraman

(H. Jayaraman)

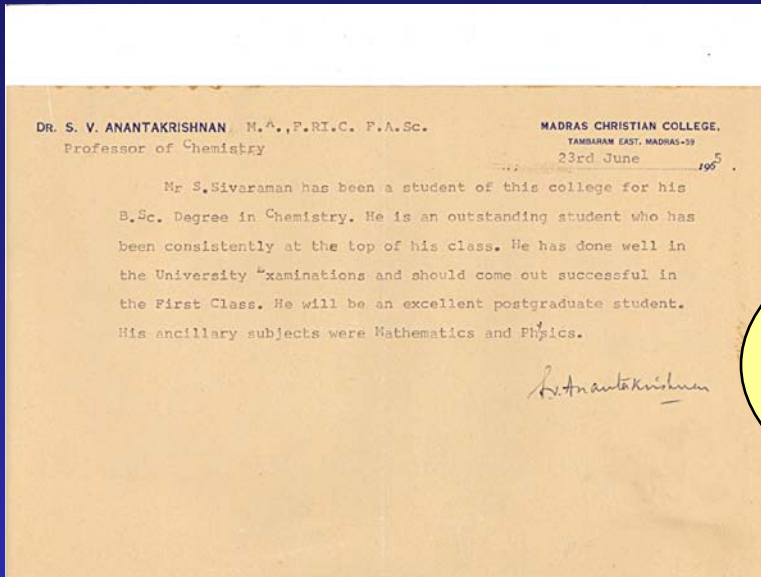
"He has been a top ranking student in his class...I have been teaching physical chemistry for the past three years and I am very much impressed with his keen sense of purpose and understanding of the subject"

H. Jayaraman

FROM MADRAS TO KANPUR

- **January 25, 1965 : The anti Hindi agitation begins all over the state of Madras**
- **Fuelled by politicians students are in the forefront of the agitation**
- **Colleges closed, examinations postponed and hostels vacated**
- **Postgraduate admissions possible only in Madras University**
- **An extraordinary intervention by SVA ; suggests that I try and seek admission to IIT Kanpur , where the Master's programme is commencing in 1965 ; tells me that IIT Kanpur has an outstanding faculty , including, two alumni of MCC – Professor M.V.George and Professor P.T. Narasimhan**
- **SVA gives me a letter of introduction to Professor P.T. Narasimhan and a letter of recommendation**

THE POWER OF THE TRUE MENTOR



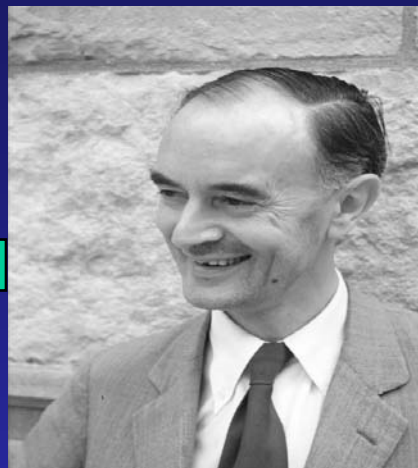
“He has been an outstanding student who has been consistently at the top of his class. He has done well in the University examination and should come out successful in the first class He will be an excellent postgraduate student”

Lesson 10 : A true mentor is one who shows you the way when you do not know where you want to go

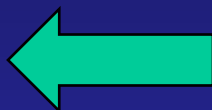
THE INGOLD CONNECTION



S.V. Ananthakrishnan
PhD with Professor Ingold
 1908-89



Christopher K. Ingold
 1893-70



*The first book
 That I bought out of
 my personal earnings
 was
 “The Structure and
 Mechanism
 In Organic
 Chemistry”, Second
 Edition, 1969, by
 C.K. Ingold*

- Ingold is considered to be the most influential chemist of the twentieth century
- The first organic chemist to use quantitative tools for the study of structure and reactivity ; the first physical organic chemist !
- Chemists use extensively the vocabulary of Ingold – nucleophilic, electrophilic, carbonium ion , R and S stereochemical notations , tautomerism, mesomerism , SN 1 and SN 2 mechanism; the use of curved arrows to indicate electron flow
- Role of steric and electronic factors in determining the reactivity of organic compounds
- Directive effects in aromatic substitution
- Kinetics of organic reactions and isotope effects to probe mechanisms

ANANTAKRISHNAN AND INGOLD CITED IN GOULD'S BOOK

My love for this book connected me to the heritage of SVA

EDWIN S. GOULD

Stanford Research Institute

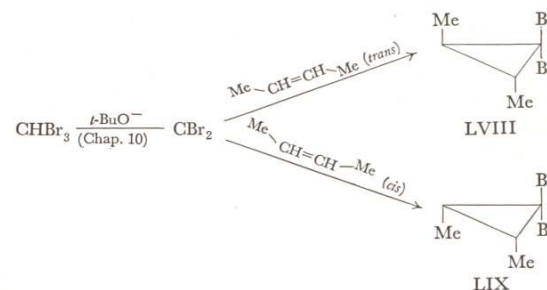
Mechanism and Structure in Organic Chemistry

This book has been purchased in the United States by the Kungur Indo-American Program with dollar funds provided by the U.S.A.I.D. for use of students at the Indian Institute of Technology Kanpur. It has been sold at a subsidised price to a student at the IITK for his personal use. Sale of the book by the student to any person or any organization other than the IITK is a breach of the terms of the original sale of the book to the student.

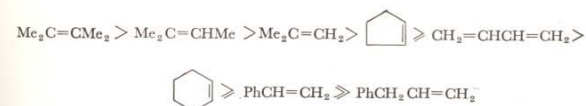
HOLT, RINEHART AND WINSTON

New York - Chicago - San Francisco
Toronto - London

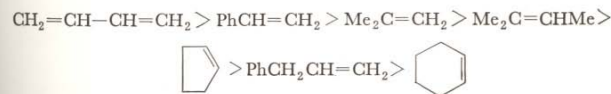
Cis Additions to the C=C Bonds. The Diels-Alder Reaction - 535



of olefins and their reactivities toward CBr_2 parallel the structure-reactivity relationships observed for heterolytic bromine addition^{42(a)} and those for olefin epoxidation.^{42(b)} The following reactivity sequence applies to all three types of addition reaction:⁴⁴



On the other hand, these olefins follow a very different sequence with respect to their reactivity toward addition of CCl_3Br in the presence of peroxides,^{42(c)} a reaction in which the C=C bond is attacked by a $\text{CCl}_3\cdot$ radical (Chap. 16). Here the following order is observed:



Thus, until more definitive evidence is available, we may infer that CBr_2 attacks as an electrophile rather than as a radical—that is, that it has zero unpaired electrons rather than two.

A somewhat similar question is associated with the familiar Diels-Alder reaction, which, the reader will recall, consists of the addition of a diene to a second unsaturated molecule—generally designated as a *dienophile* (which often, but not invariably, has one or more electron-attracting groups in conjugation

⁴² (a) Ingold, *et al.*, *J. Chem. Soc.*, 1931, 2354; 1935, 984, 1396. (b) Swern, *J. Am. Chem. Soc.*, 69, 1692 (1947). (c) Kharasch, *et al.*, *J. Org. Chem.*, 14, 239, 537 (1949); 18, 328 (1953).

PAPERS BY ANANTAKRISHNAN AND INGOLD

(from the Ph D thesis of SVA)

Addition of bromine to ethylene and substituted ethylenes : Role of methyl substituent on reactivity

984 Anantakrishnan and Ingold : Influence of Substituents on the

226. Influence of Substituents on the Additive Reactivity of Ethylene Derivatives. Part II. Effects of Catalysts on the Relative Rates of Addition of Bromine.

By S. V. ANANTAKRISHNAN and CHRISTOPHER K. INGOLD.

THE object of these researches is to determine, first, the effect of a single group, R, on the rate of reaction of RCH_2CH_2 towards addenda, and secondly, the rules under which such effects are compounded when several R's are simultaneously attached to the group C_2C . When the electronic classification (nucleophilic or electrophilic) of the reagent is known, the influence on reaction rate should be capable of correlation with the modes of electron displacement characteristic of R.

Part I (Ingold and Ingold, J., 1931, 2354) commenced a study, which this paper continues, of the reactivity of ethylene derivatives towards the unambiguously electrophilic reagent, bromine. It was shown that the effect of substituents, R, could not in general be deduced by analogy with other reactions (tautomerism, aromatic substitution, etc.), because the polarisation and polarisability of a group contribute with different relative weights for every reaction, and it is impossible to tell in advance how these factors should be weighted in the addition reaction with bromine. It was also shown that the intrinsic effects of R's when present singly could not in general be deduced by an analysis of results for compounds in which several R's are present together: two activating groups could jointly deactivate, and vice versa. Theory defines the scope of these ambiguities, and it happens that most known results are affected by them. Therefore the authors of Part I sought to establish the fundamental relationships concerning group influence in the bromine addition reaction by choice of the proper examples, which, however, are not convenient for the purpose of detailed kinetic measurement. Accordingly, recourse was had to the com-

J. Chem. Soc, 1935

1396 Anantakrishnan and Ingold : Influence of Substituents on the

334. Influence of Substituents on the Additive Reactivity of Ethylene Derivatives. Part III. Further Experiments on the Addition of Bromine in Solution.

By S. V. ANANTAKRISHNAN and CHRISTOPHER K. INGOLD.

In this paper we report an extension of the investigation commenced in Part I (Ingold and Ingold, J., 1931, 2354) on the relative rates of addition of simple ethylene derivatives towards the electrophilic reagent, bromine. We have throughout standardised the catalytic conditions for this reaction by the method worked out in Part II (Anantakrishnan and Ingold, this vol., p. 334), a sufficient quantity of hydrogen bromide being added initially to cover up completely the autocatalysis of the reaction as well as any adventitious catalysis.

It was shown in Part I that the intrinsic effect of a given substituent on the reactivity of an ethylene bond towards a reagent such as bromine can be deduced only by reference to certain structural types. These may contain either just the single substituent under consideration or combinations of substituents subject to definite restrictions; for many combinations give rise to mutual effects which seriously distort, and may even invert, the separate effects of the groups. Theory, confirmed by observation as far as these have been extended, shows when it is possible for this to happen, and therefore indicates the structural types which may most profitably be studied.

Within such limits we have extended the work of Part I on the methyl and phenyl substituents, and on the combination of methyl groups. We have also studied the aldehydic group and its combination with methyl. The experiments were carried out at -78° , and, as before, in acetylene chloride solution with exclusion of light. The method was to allow two ethenic substances to compete for a deficit of bromine, and to calculate the ratio of the rate constants of the two reactions from the composition of the mixture of disubstituted bromide after absorption of the whole of the bromine. It was shown in Part II that the ratios thus obtained are insensitive to the concentrations of the reactants, to the concentration of the hydrogen bromide catalyst, and, within considerable limits, to the temperature. By means of a suitable set of such ratios it is possible to obtain "reduced rates", i.e., rates of reaction for ethylene derivatives expressed as multiples or fractions of the rate for ethylene itself.

Table I summarises the experiments carried out in the previous paper and in this paper, together with the corresponding ratios of the rate constants and the calculated reduced rates. The tabulated figures are based on the observational data in Table II on p. 1399.

Expt. No.		X	Y	k_1/k_2	Reduced rate for Y	Mean
15-19	CH ₃ CH ₂	CH ₃ CH ₂	CH ₃ CH ₂	2.20	2.20	2.20
20-24	CH ₃ CH ₂	CH ₃ CH ₂	CH ₃ CH ₂	2.47	2.47	2.47
25	CH ₃ CH ₂	CH ₃ CH ₂	CH ₃ CH ₂	2.22	$0.43 \times 2.20 = 2.00$	2.00
26	CH ₃ CH ₂	CH ₃ CH ₂	CH ₃ CH ₂	2.50	$0.33 \times 2.50 = 2.82$	2.82
27-31	CH ₃ CH ₂	CH ₃ CH ₂	CH ₃ CH ₂	2.9	2.9	2.9
32-36	CH ₃ CH ₂	CH ₃ CH ₂	CH ₃ CH ₂	2.5	2.5	2.5
37-41	CH ₃ CH ₂	CH ₃ CH ₂	CH ₃ CH ₂	10.4	10.4	10.4
42-46	CH ₃ CH ₂	CH ₃ CH ₂	CH ₃ CH ₂	12.6	12.6	12.6
47-51	CH ₃ CH ₂	CH ₃ CH ₂	CH ₃ CH ₂	2.54	$2.54 \times 0.50 = 1.27$	1.27
52-56	CH ₃ CH ₂	CH ₃ CH ₂	CH ₃ CH ₂	2.54	2.54	2.54
57-61	CH ₃ CH ₂	CH ₃ CH ₂	CH ₃ CH ₂	0.13	$0.13 \times 2.20 = 0.29$	0.29
62-66	CH ₃ CH ₂	CH ₃ CH ₂	CH ₃ CH ₂	2.2	2.2	2.2
67-71	CH ₃ CH ₂	CH ₃ CH ₂	CH ₃ CH ₂	2.2	$1.0 \times 2.2 = 2.2$	2.2

In spite of the new data, nothing need be added to the discussion in Part I of the effects caused by the groups Ph, Et, and CO₂H. The following remarks will therefore be confined to the effects of the substituents Me and CH₂.

The methyl group releases electrons and therefore should facilitate the addition of electrophilic reagents such as bromine. Thus the reduced velocity for propylene should

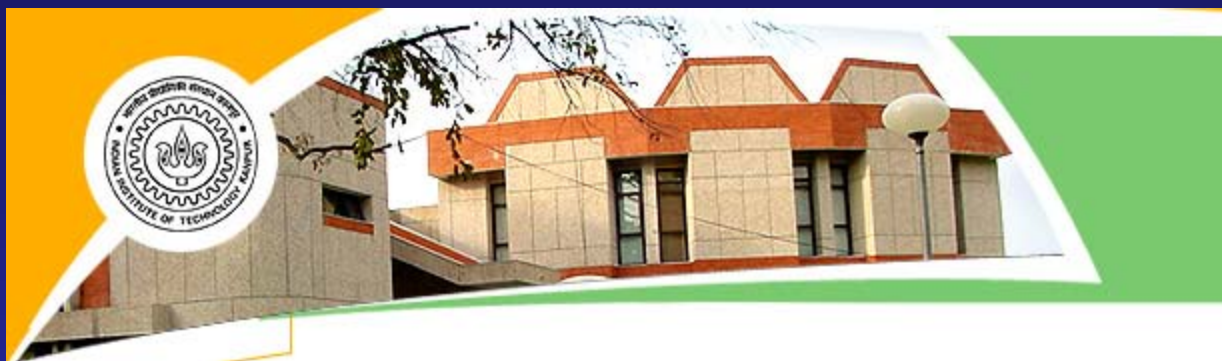
Lesson 11 : Knowledge is akin to insurance; You never know when you will need it.

DEPARTMENT OF CHEMISTRY, IIT KANPUR (1965-67)

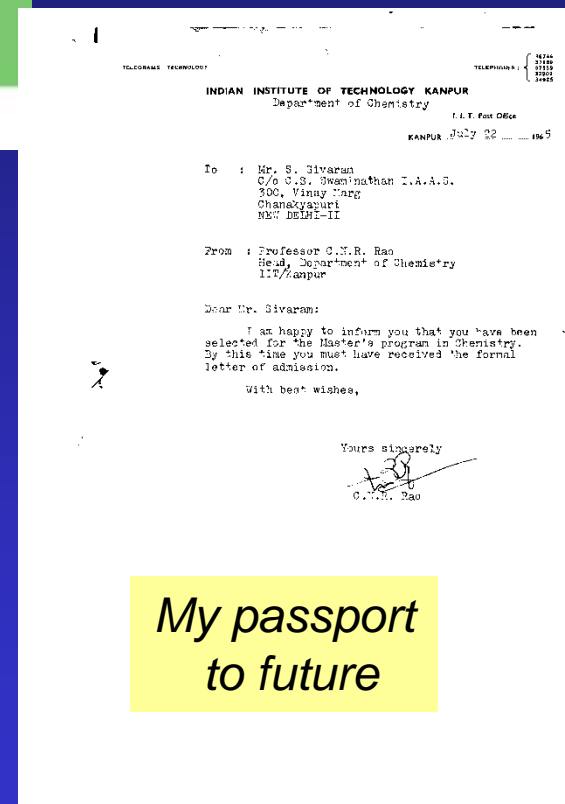


- From an old institution to a new institution
- A class of seven – first batch of MSc in IIT
- Extraordinary faculty with exceptional pedigree (Prof Rao – Kenneth Pitzer, Professor Narasimhan – Karplus, Professor George – Gilman, Barton and Huisgen, Professor Ranganathan- Woodward, Professor Chakravorti – Cotton etc. The universities they came from was astounding, Harvard, Berkeley, Purdue, Columbia, Munich, MIT, Imperial College , London !

DEPARTMENT OF CHEMISTRY, IIT KANPUR (1965-67)



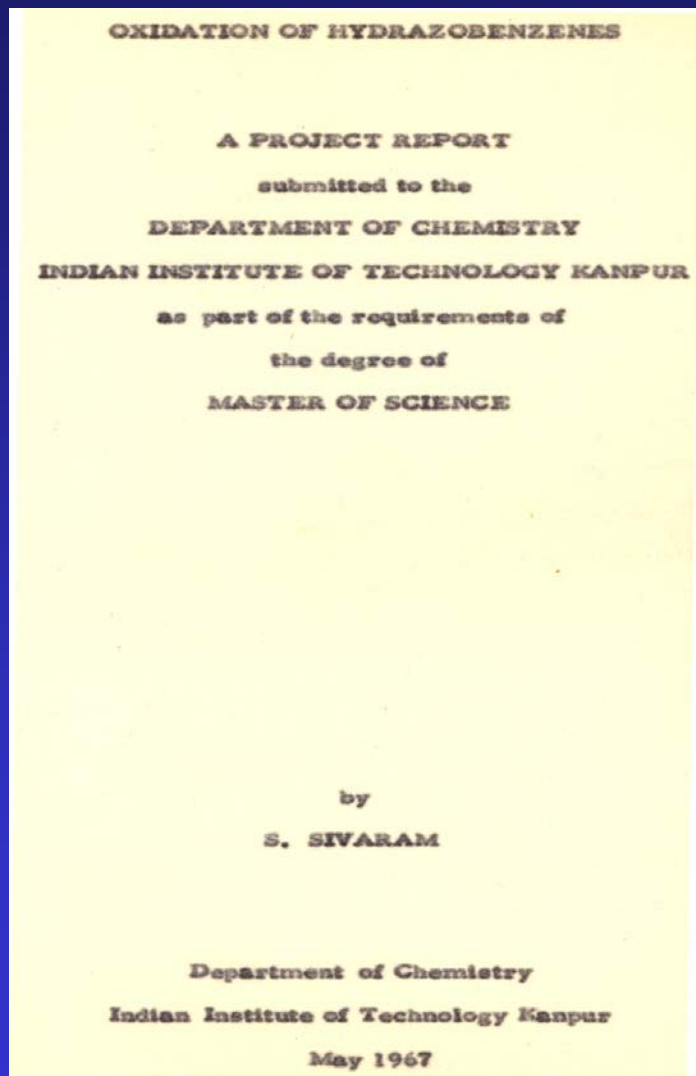
- Research in the air; Frequent seminars and lectures; lights burning in the lab late in the night !
- A large number of Ph D students – feel elevated in their company
- Continuous evaluation; surprise quizzes; no final examinations
- Over thirty books given free by US AID; The beginning of the personal library
- Faculty exceedingly informal and accessible
- Summer and M. Sc research in Professor George's laboratory in 1965



*My passport
to future*

Lesson 12 : Teach less; encourage self learning; make learning a pleasure

THE FIRST TASTE OF RESEARCH



- Oxidation of hydrazobenzenes to azobenzenes using radical anions
- Na and K Naphthalene as radical anion source
- Effect of para substituent on the rate of oxidation
- Rate measured using a Beckman DU UV-Visible Spectrophotometer
- My little thesis undergoes about ten revisions in the hands of Professor M. V. George !

Lesson 13 : Pay attention to details; small things are important in science

A MENTOR SETS THE COURSE (once again!)

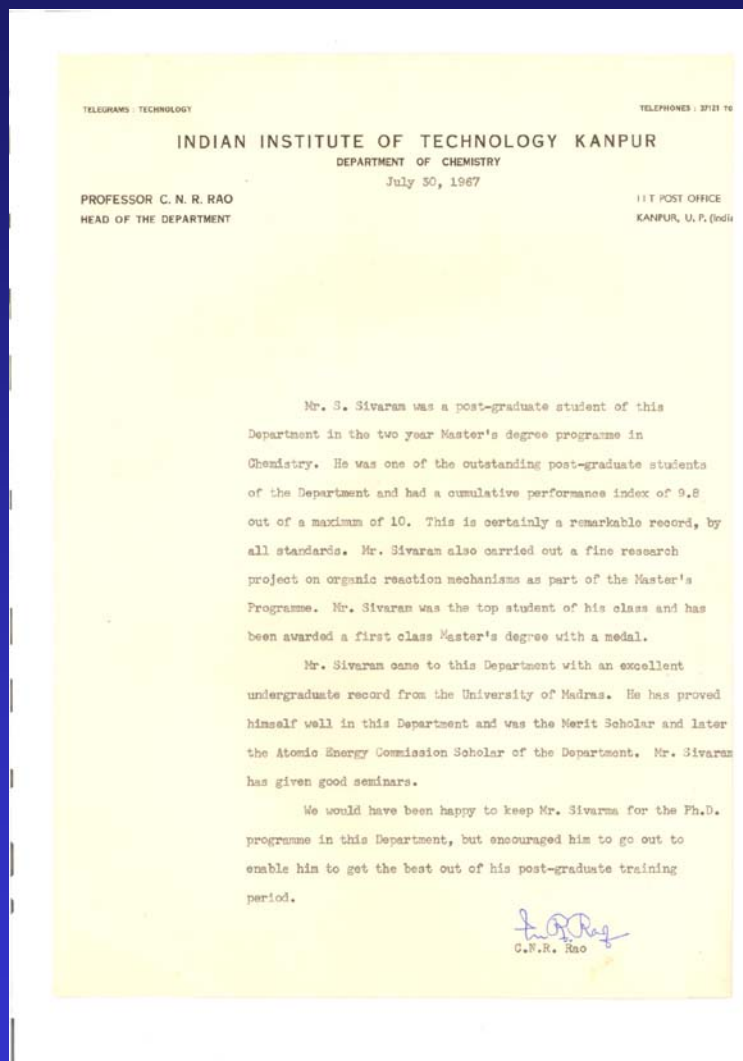


Professor C. N. R. Rao

- In June 1967, Professor Rao asks me what I propose to do after M.Sc
- He suggests I go abroad for doing my Ph. D
- When I say that I am interested in Organic Chemistry he suggests that I work with Professor H.C Brown at Purdue University
- He writes a letter on my behalf to Professor Brown ; I receive a letter of appointment as a fully paid research assistant from Professor Brown in July 1967

Lesson 14 : A true mentor is one who is more concerned about his student, not himself

A LETTER FROM PROFESSOR RAO



“He was an outstanding postgraduate student of the department and had a cumulative performance index of 9.8 out of 10. This is certainly a remarkable record by all standards.....We would have been happy to keep Sivaram for the Ph D programme in this department, but encouraged him to go out to enable him to get the best out of his post graduate training period “

***Professor C. N.R. Rao, F.R.S
July 30, 1967***

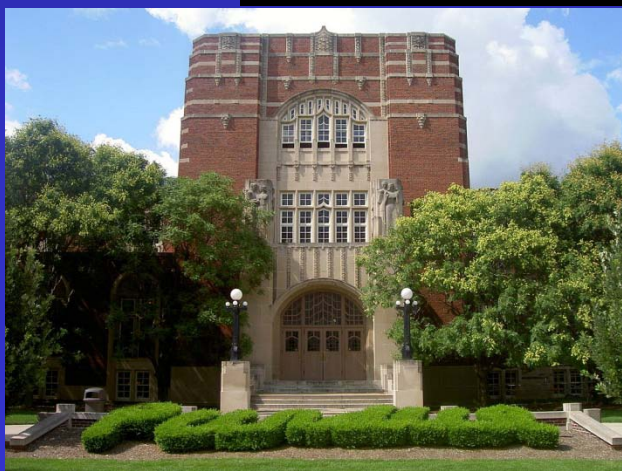


FROM KANPUR, INDIA TO W.LAFAYETTE, INDIANA, USA

August 1967



Purdue University



**R. B. Wetherill
Laboratories
for Chemistry
An ACS
National
Historic
Landmark**





PURDUE UNIVERSITY, DEPARTMENT OF CHEMISTRY, 1967-71

- Professor Brown gives me a choice of two topics : Organoborane chemistry and carbonium ion chemistry; I choose the latter
- The die is cast . I am now on my way to becoming a physical organic chemist !
- Rich experience; learning to be independent in thought and action
- Head days, great teachers (Nathan Kornblum, Robert Benkeser, Derek Davenport, Fred Mclafferty, Harry Morrison, Henry Feuer)
- Research and classroom learning
- Great friendships

BRIDGED PHENONIUM ION OR EQUILIBRATING CARBONIUM ION ?

xlix

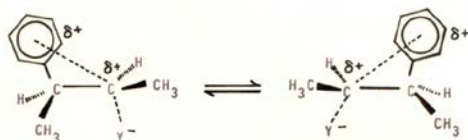


Figure 2. A Possible k_A Intermediate

lxxvi

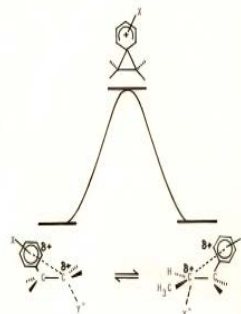
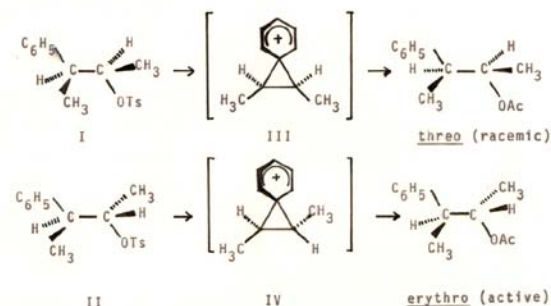


Figure 1. A Possible Intermediate in the Aryl Assisted Pathway

4

active products as shown in scheme 2.

Scheme 2

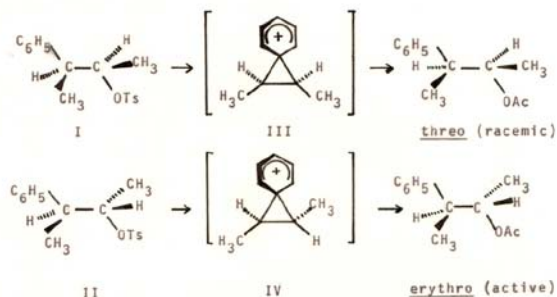


The initial simplicity of the phenonium ion theory, however, was largely lost when results of more detailed study on the rates and products became available. It was observed that the rate of racemization was considerably greater than the rate of acetolysis. This was attributed to ion pair return by Winstein and Schreiber.¹⁰ The intermediacy of phenonium tosylate ion pairs were invoked to explain this observation. These ion pairs return to starting material as well as dissociate to products, all processes occurring with

RETENTION IN STEREOCHEMISTRY IS INCONSISTENT WITH A SYMMETRIC INTERMEDIATE

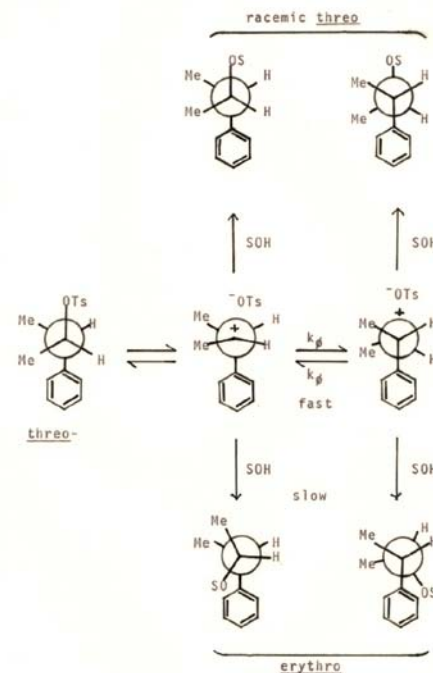
active products as shown in scheme 2.

Scheme 2

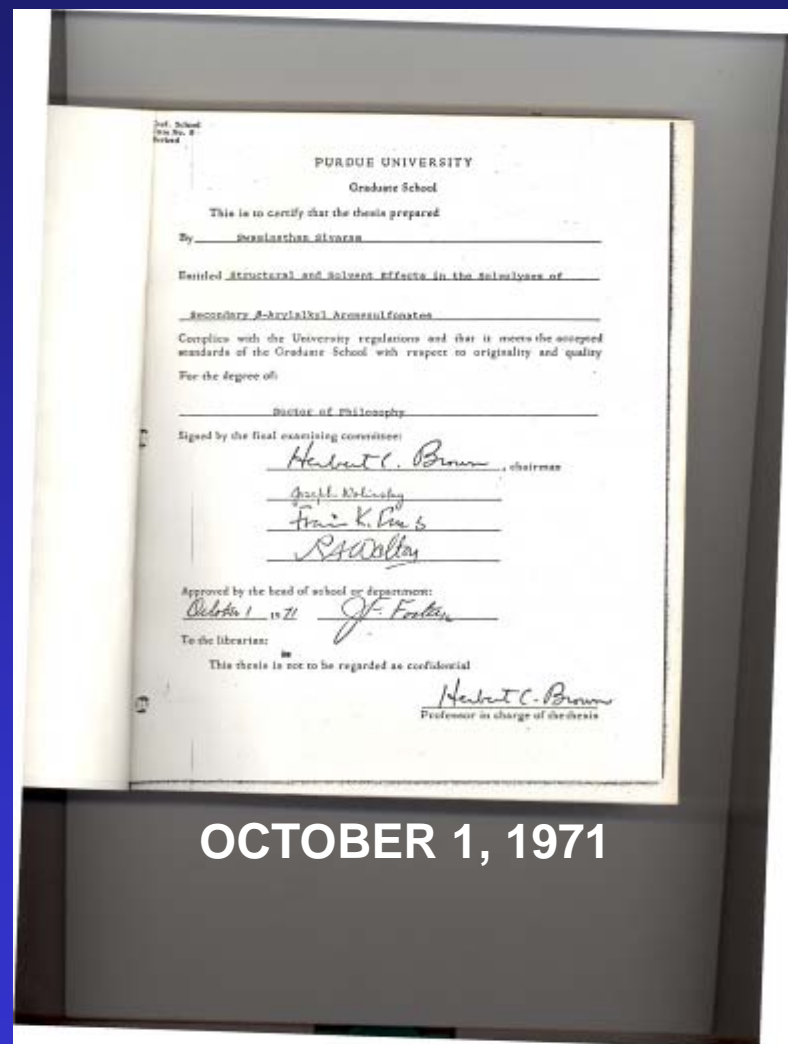
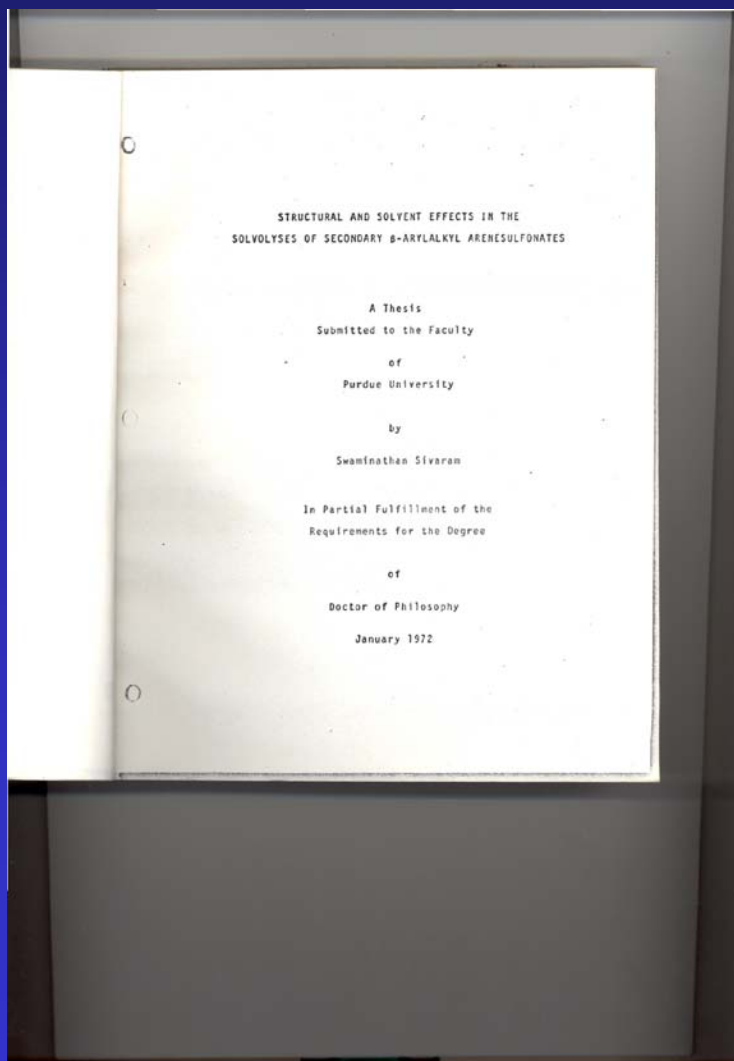


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Scheme 2

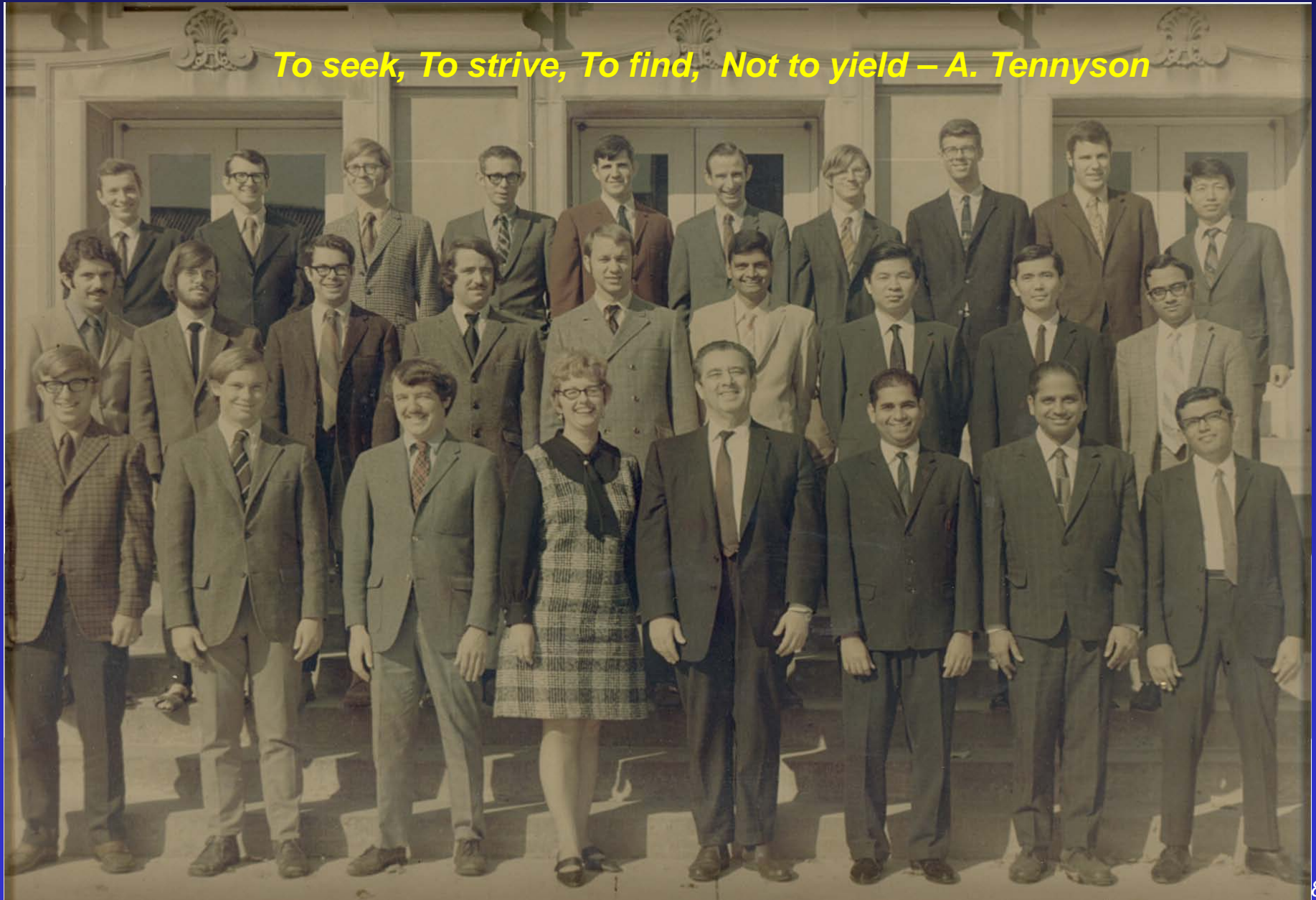


STRUCTURAL AND SOLVENT EFFECTS ON THE SOLVOLYSIS OF SECONDARY β -ARYLALKYL ARENESULFONATES



The Brown Research Group 1970

To seek, To strive, To find, Not to yield – A. Tennyson



ANOTHER LETTER THAT SETS ME ON A NEW JOURNEY

PURDUE UNIVERSITY
DEPARTMENT OF CHEMISTRY
LAFAYETTE INDIANA 47907

November 12, 1970

Professor Joseph M. Kennedy
Department of Chemistry
University of Akron
Akron, Ohio 44304

Dear Joe:

It is a long time since we had an opportunity to get together. I hope that you are getting nicely settled at Akron and are enjoying the academic life as much as you had hoped.

As you know, we have been doing considerable research on carbonium ions generated in solvolytic reactions. However, I felt that my students should be familiar with the growing amount of work being done on carbonium ion intermediates in non-solvolytic reactions. Accordingly, I instituted a program in our research seminars of reviewing Olah's work on carbonium ions in NMR solvents, George Kramer's work on trapping ions by hydride transfer, and your own work on polymerization.

One of my students, S. Sivaram, currently studying the 3-phenyl-2-butyl cation, has become very interested in this area. He should be completing his research for the Ph.D. degree sometime next summer. He is interested in a postdoctorate appointment which will extend his training to this new area. Accordingly, I recommended that he write you.

He is an exceptional young man. Scholastically, he ranks in the upper 10% of our students. He is industrious and energetic, and works very well independently, with a minimum of guidance. He is exceptionally pleasant personally, and a pleasure to work with.

If you can accommodate him in your research group next academic year, I am confident you would find him to be an exceptionally productive coworker and one with whom you would enjoy working.

Sincerely yours,

Herbert C. Brown

HCB:aw

bcc: S. Sivaram -

“One of my students Sivaram who is studying the 3-phenyl-2-butyl cation has become interested in your work on carbocationic polymerization. He should be completing his Ph D degree sometime this summer. Accordingly I recommended that he write to you.

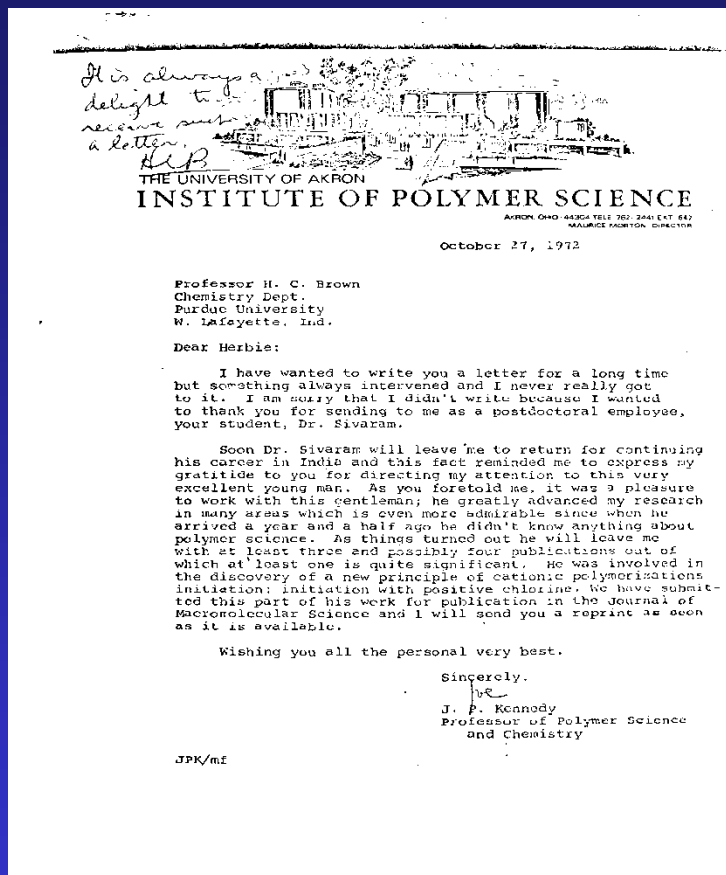
He is an exceptional young man. Scholastically he ranks in the upper 10 % of our students. He is industrious and energetic and works well independently with a minimum of supervision

I am confident that you will find him to be an exceptionally productive coworker and one with whom you will enjoy working “

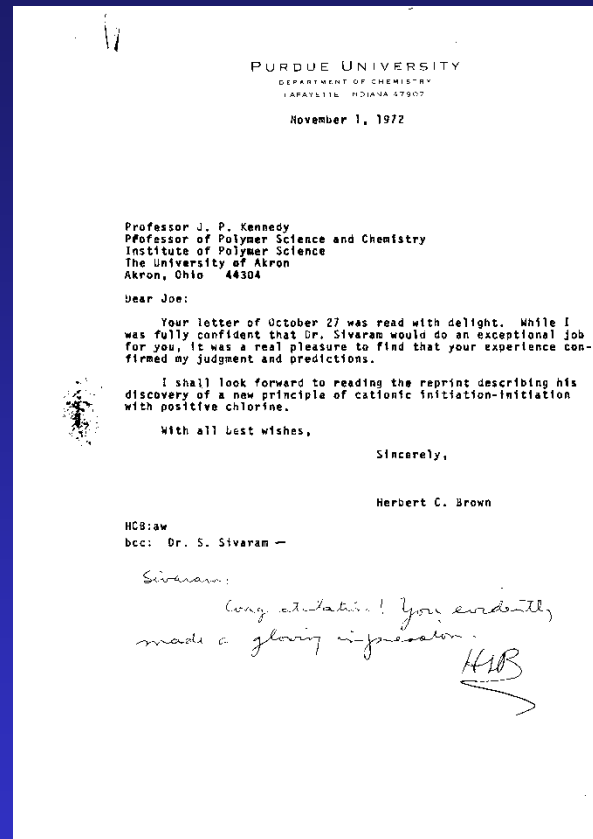
***Professor H.C.Brown
November 12, 1970***

LETTER FROM PROFESSOR KENNEDY TO PROFESSOR BROWN

OCTOBER 27, 1972

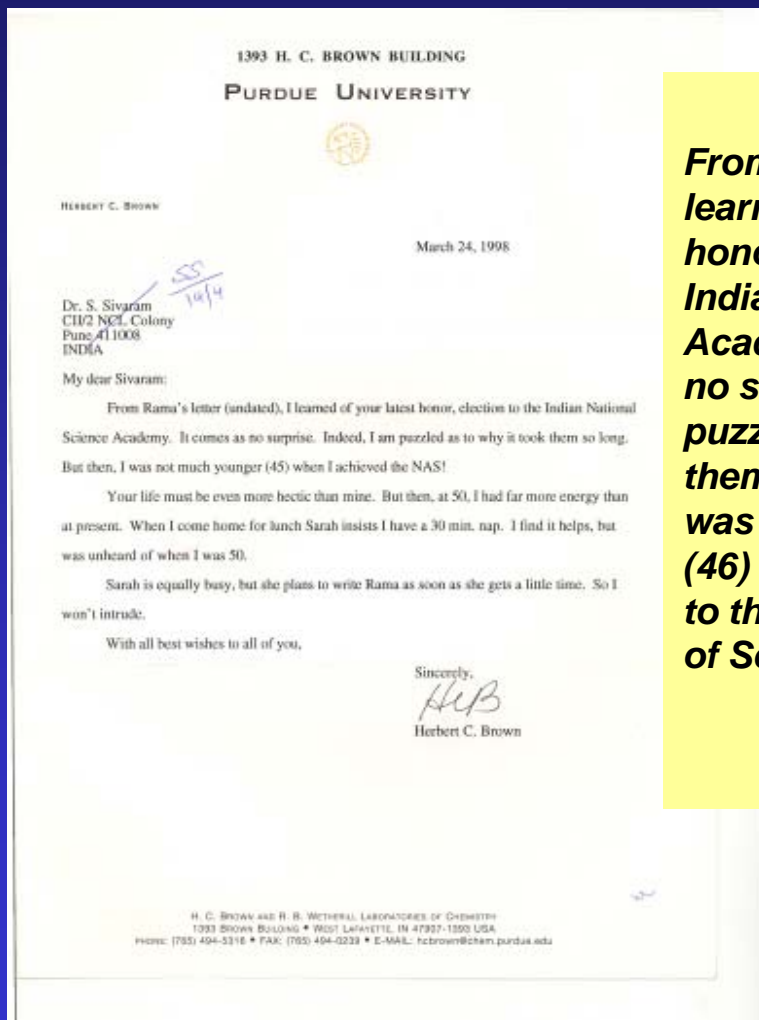


“While I was fully confident that Sivaram would do an exceptional job for you it was a real pleasure to find out that your experience confirmed my judgment and predictions”



Lesson 15 : There is no greater reward for a student than to know that he has lived up to his teacher's expectations

LETTER FROM PROFESSOR BROWN DATED MARCH 24, 1998 AND E MAIL DATED JUNE 10, 1998



From Rama's letter I learnt of your latest honor, election to the Indian National Science Academy. It comes as no surprise. I am puzzled as to why it took them so long. But then, I was not much younger (46) when I was elected to the National Academy of Sciences !

Congratulations!

Received: from CV3.CHEM.PURDUE.EDU (cv3.chem.purdue.edu [128.210.43.119]) by ems.ncsl.res.in with SMTP (8.7.1/8.7.1) id CAA25538 for <sivaram@ems.ncsl.res.in> Thu, 11 Jun 1998 02:22:02 +0530 (IST)
Date: Thu, 11 Jun 1998 02:22:02 +0530 (IST)
Received: from [128.210.43.27] by CV3.CHEM.PURDUE.EDU with SMTP; Wed, 10 Jun 1998 (5:51:25 -0500 EST)
Message-ID: <031107001a44c020b02a@128.210.43.27>
Mime-Version: 1.0
Content-Type: text/plain; charset="us-ascii"
To: sivaram@ems.ncsl.res.in
From: "H. C. Brown" <hcbrown@chem.purdue.edu>
Subject: Congratulations!
X-1 IDL: 897536418.008
X-Mozilla-Status: 8000

By GUPTA SIVARAM:

It is a rare experience to have a former student return to India and do research which attracts the attention of the academic community abroad. It is an even rarer experience for such research to attract the attention of industrial laboratories and organizations here to the point that joint research projects are initiated. Congratulations!

I shall look forward to seeing you on one of your upcoming trips to the US. Incidentally, if you can manage to bring your wife along, Sarah would be exceptionally delighted. On your last visit Sarah enjoyed her company enormously.

With all best wishes, sincerely,

Herbert C. Brown

Prof. Herbert C. Brown
H. C. Brown and R. B. Wetherill
Laboratories of Chemistry
Purdue University
West Lafayette, Indiana 47907 USA
E-MAIL: hcbrown@chem.purdue.edu
FAX: 01-317-494-0229

It is a rare experience to have a former student return to India and do research which attracts the attention of the academic community abroad. It is an even rarer experience for such research to attract the attention of industrial research laboratories here to the point that joint research projects are initiated. Congratulations !

Lesson 16 : A true mentor is one who rejoices in the success of his student



I met Professor Brown for the last time in 2002 on his 90th Birthday ; he was visibly and genuinely proud that I had been appointed Director, NCL



WISHING YOU A
World of Peace
2001

HERB AND SARAH BROWN

From HCB, SBB, Tamar
and Ronni (granddaughters)
Celebrating HCB's 90th



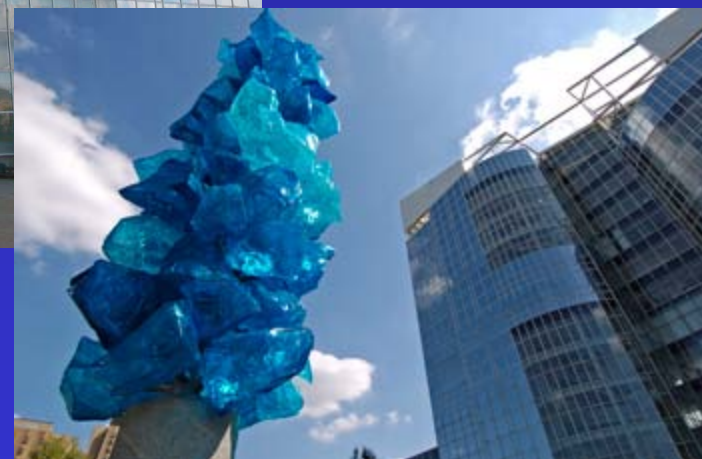
CARBOCATIONIC POLYMERIZATION AT THE UNIVERSITY OF AKRON, INSTITUTE OF POLYMER SCIENCE (1971-73)

- Polymerization of isobutene using cationic initiators
- Catalyzed addition of halogens to isobutylene for initiation of polymerization
- Mechanism of alkylation of t-butyl halides with alkylaluminums
- Alkylation of olefins with HCl / trimethylaluminum
- An opportunity to learn polymer science, both theory and practice
- Understand how chemistry becomes useful when used to make materials

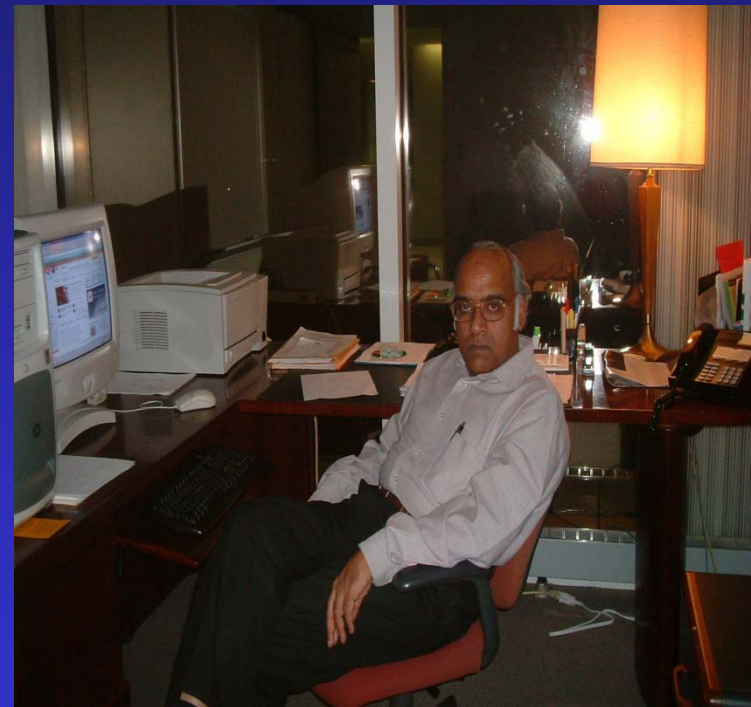
“It is exhilarating to explore, discover and understand.....but it is no longer sufficient; chemistry not only has to be elegant but must also be useful”

Professor J.P.Kennedy

MEMORIES RELIVED : HAROLD A. MORTON DISTINGUISHED PROFESSOR AT THE UNIVERSITY OF AKRON IN FALL 2006



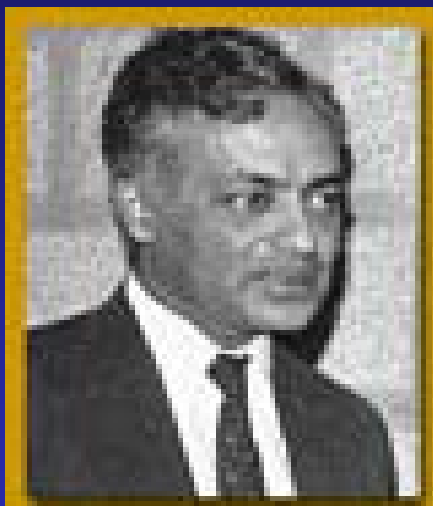
COLLEGE OF POLYMER SCIENCE AND ENGINEERING, THE UNIVERSITY OF AKRON (Fall 2006)



PROFESSOR KENNEDY AND INGRID KENNEDY, AKRON, NOVEMBER 2006



RETURN TO INDIA : TO INDIAN PETROCHEMICALS CORPORATION LIMITED, BARODA (1973-88)



Dr S. Varadarajan
CMD, 1974-80
Taught me the principles of management and how to think big

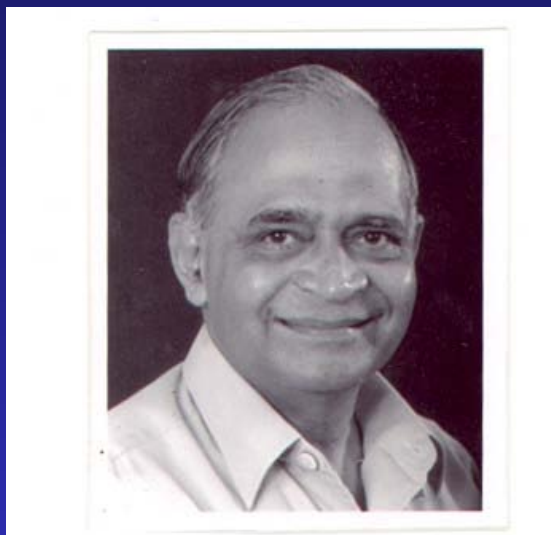
- A career in industry
- Saw chemistry in action
- Established a research laboratory
- Initiated research in the area of Ziegler Natta Catalysts for Polyolefins
- Part of a team that developed a process for acrylic esters from acrylonitrile in collaboration with NCL
- Participated in the complete chain consisting of process development, process design, basic engineering and through commissioning of a 10,000 tpa plant at Baroda
- Participated in several technology acquisition activities



Dr P.G. Menon
Manager (R&D)
1973-1979
Initiated me into industrial R&D and taught me that no problem is too small to solve

ANOTHER INSPIRATIONAL MENTOR

- Met him for the first time in 1973 in his crowded and modest office at the Department of Chemical Engineering, UDCT, Matunga, Bombay
- At a young age I was struck by his incredible breadth of knowledge, memory for facts, erudition and enthusiasm.
- Kindled my interests and taught valuable lessons in industrial chemistry, petrochemicals and catalysis; taught me the how to understand and value of technology
- Fortunate to have received his continuous mentorship for over four decades



Professor M. M. Sharma

The most important lesson learnt

- Academic excellence and application of science in industry are not mutually exclusive***
- Knowledge in depth and breadth is essential for becoming a successful professional***



AN INVITATION TO JOIN NCL



Dr R. A. Mashelkar

11/11/86

My dear Sir,

I received your letter a week back but decided to delay my response until I had discussed the matter fully with Dr. Doraiswamy. I have done this now and I am pleased to you with confidence.

First of all let me say that your decision to join NCL has one of the best reasons - that in recent years you have been more & more regarded as a scientist and a person and I could not have thought of a better person to lead our efforts in Polymer Chemistry. LKD too has overjoyed to hear about your decision.

I appreciate your problems about the possibilities of NCL. You have been with the department for a long time and any new arrangement implies building up of a new team which cannot be snapped suddenly and with immediate effect. LKD & I appreciate the problem.

LKD is going to speak to Dr. Gunguly, who he sees him in the next board meeting. Knowing LKD, I expect him to handle it most sensitively. LKD is sure that your join is as soon as possible. He told me that he will advise Dr. Gunguly that your association with the Polymer project could be kept going even if you join NCL. Let us see how the Gunguly reacts to it.

I was in Delhi during the last week here to Director's Conference. I met Dr. SV during this time. I had a detailed discussion with him about the dilemma you are facing. You know what his

1

remedy, here when I mentioned to him about Dr. Gunguly's suggestion of joining NCL after 3 to 4 years. He said - 'It's 4 years but is too late for Siram!' That is my own opinion too.

Well, that I have summarized above, Siram, in what LKD, SV and I think about the changeover. NCL can get very better in years to come. Together, we can really build our institute where the highest scholarly traditions are followed and which will be the envy of the rest of the world. NCL is already highly regarded for I am sure (just as you are) of the volume, the gaps and the problems. The greater challenge & opportunity awaits for all of us. We can do it together.

I still regard to you as soon as LKD speaks to Dr. Gunguly.

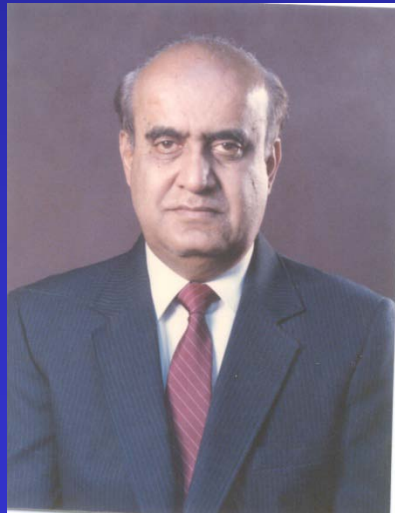
With my warmest personal regards to yourself, Rupa & love to the children,

Yours sincerely,
Rash

B

One part of S.F. will have to be addressed. With the school spent in which days more, I expect the whole thing to have at least 6 to 7 weeks.

Ram



Dr. L.K. Doraiswamy

A CALL ONE CAN HARDLY REFUSE

- *“ NCL can only get better in the years to come. Together we can really build an institute where the highest traditions of scholarship are followed and which will be the envy of rest of the world. NCL is already highly regarded but I am aware of the lacunae, the gaps and the problems. The greatest challenge and opportunity awaits all of us. We can do it together ”*

Dr R.A. Mashelkar F.R.S

Letter dated 11 November 1986



- Leadership of Polymer Chemistry Division
- An opportunity to build the Division – Define programmes and philosophy
- Build a personal research program with PhD students – Mentoring
- NCL is a very competitive environment; only the best survive ! Highly decorated scientists
- Culture of patenting
- Collaboration with industry : Use new or known science to solve problems
- Opportunities for creating new science

A remarkable ambience for productive science; respect for scholarship and a distinguished lineage of remarkable leaders

FUNCTIONAL POLYMERS THROUGH CONTROLLED CHAIN GROWTH POLYMERIZATION

- Functional initiators
 - *Anionic, cationic, free radical, GTP, ROP*
- Functional monomers
 - Free radical, GTP
- Protected functional monomers
 - Anionic, GTP, metal catalyzed polymerization
- Functional termination of living chain ends
 - Anionic, GTP, cationic, free radical
- Controlled catalytic chain transfer
 - Free radical, *metal catalyzed polymerization*

POLYESTERS AND POLYCARBONATES: SYNTHESIS, STRUCTURE AND PROPERTIES

- Building diversity in aromatic polyesters : Structure and morphology
- Polycarbonates and co-polycarbonates : Synthesis via solid state polymerization
- Fully aliphatic polyesters as biodegradable polymers : Synthesis, structure and properties

FUNCTIONAL POLYOLEFINS

IN-CHAIN FUNCTIONALIZATION

- Copolymerization with a functional monomer
- Copolymerization with a precursor monomer with masked functionality which does not interfere in copolymerization

CHAIN-END FUNCTIONALIZATION

- Introduction of functional groups by chemical reaction on preformed polymer

NATIONAL CHEMICAL LABORATORY (2002- present)

- **Challenge of management of one of the largest publicly funded laboratories in India**
- **Charting the future of the Laboratory**
- **Mentoring and nurturing younger scientists and students**
- **Resource management**
- **Building the pipeline for emerging science and technology**
- **Influence S&T policy of India**

SOME MOMENTS TO CHERISH



*Dr Mashelkar, President ,
presenting the
Vishwakarma Medal. Indian
National Science Academy,
2006*



*Professor Rao presenting the
Silver Medal of the Chemical
Research Society of India ,
2002*



SOME MOMENTS TO CHERISH

President Abdul Kalam conferring Padma Shri , March 2006



LESSONS THAT I HAVE LEARNT

- We rarely get what we want ; always make the best of second choice
- Provide early opportunities to learn leadership and organizational skills
- Learning outside the classroom is more important than inside the classroom
- Be generous with praise ; It does wonders
- However big you are, show that you care for everyone in your institution
- Teach to appreciate the beauty of the written word. Create the love for reading. Language is the window to the soul
- If you want to find out a person's real aptitude, give him a gift and ask him to select a book from a bookstore
- Teachers who are committed and show genuine interest in their pupil make good institutions great

LESSONS THAT I HAVE LEARNT

- A liberal education is far more important than learning a few subjects; You can rebuild a façade but can lay the foundation only once
- A true mentor is one who shows you the way when you do not know where you want to go
- Knowledge is akin to insurance; You never know when you will need it
- Teach less; encourage self learning; make learning a pleasure
- Pay attention to details; small things are important in science
- A true mentor is one who is more concerned about his student, not himself
- There is no greater reward for a student than to know that he has lived up to his teacher's expectations
- A true mentor is one who rejoices in the success of his student

LIFE'S LESSONS

- The power of the mentors - encounters with great minds who see farther than you do
- Being at the right place at the right time; an opportunity to be associated with great institutions
- Building a robust foundation
- Liberal education that teaches you to keep your mind open and observe your environment critically
- Repeated learning and relearning experiences
- An opportunity to build both character and competence

***In a life's journey, every encounter is a matter of chance
What great minds and institutions teach you is to stand tall, look at
the skies, dream passionately and relentlessly work to convert
dream into reality***

**The moving finger writes; and having writ,
 Moves on : nor all your piety nor wit
 Shall lure it back to cancel half a line
 Nor all your tears wash out a word of it**

***Fitzgerald's Translation
 Rubaiyat of Omar Khayyam***

THANK YOU

